

1 ALAN M. CAPLAN (SBN 49315)
2 APRIL M. STRAUSS, Of Counsel (SBN 163327)
3 BUSHNELL, CAPLAN & FIELDING, LLP
4 221 Pine Street, Suite 600
5 San Francisco, CA 94104
6 Telephone: (415) 217-3800
7 Facsimile: (415) 217-3820

8 Attorneys for Plaintiff ENVIRONMENTAL LAW FOUNDATION
9 (Additional Attorneys on Signature Page)

10 SUPERIOR COURT OF THE STATE OF CALIFORNIA
11 COUNTY OF SAN FRANCISCO, UNLIMITED JURISDICTION

12 IN RE VINEGAR LITIGATION

13 CASE NO. CGC-03-421108
14 (consolidated with Nos. CGC-04-428945
15 and CGC-04-435440)

16 CONSENT JUDGMENT AS TO
17 DEFENDANTS BERBERIAN
18 ENTERPRISES, INC. AND BRISTOL
19 FARMS; ORDER

1 1 INTRODUCTION

2 1.1 October 13, 2004, the Environmental Law Foundation, individually and on
3 behalf of the general public ("ELF") filed a Complaint for civil penalties, restitution and
4 injunctive relief ("Complaint") in San Francisco County Superior Court ("Action").
5 Berberian Enterprises, Inc. ("Berberian") and Bristol Farms, two of the defendants in the
6 Action, shall hereinafter be referred to as "Settling Defendants."

7 1.2 Settling Defendants are a corporation that employ more than ten persons and
8 sells Wine Vinegars to persons in the State of California. For purposes of this Consent
9 Judgment, the term "Wine Vinegar" shall have the meaning set forth in section 6.2.

10 1.3 ELF's Complaint alleges that the Settling Defendants manufactured,
11 distributed and/or sold Wine Vinegar containing lead in an amount that resulted in an
12 exposure to consumers in violation of the provisions of the Safe Drinking Water and Toxic
13 Enforcement Act of 1986 and Health & Safety Code §§ 25249.5, *et seq.* (Proposition 65),
14 and Business & Professions Code §§ 17200, *et seq.* ("Unfair Competition Law"), by
15 knowingly and intentionally exposing persons to a chemical known to the State of California
16 to cause reproductive toxicity, namely lead, without first providing a clear and reasonable
17 warning to such individuals.

18 1.4 For purposes of this Consent Judgment only, the parties stipulate that this
19 Court has jurisdiction over allegations of violations contained in the Complaint and personal
20 jurisdiction over the Settling Defendants as to the acts alleged in the Complaint, that venue
21 is proper in the County of San Francisco and that this Court has jurisdiction to enter this
22 Consent Judgment as a resolution of all claims which could have been raised in the
23 Complaint based on the facts alleged therein.

24 1.5 Settling Defendants deny, generally and specifically, the allegations set forth
25 in the Complaint.

26 1.6 For the purpose of avoiding prolonged litigation, the parties enter into this
27 Consent Judgment as a full settlement of all claims that were raised in the Complaint based
28 on the facts alleged therein, or which could have been raised in the Complaint arising out of
the facts alleged therein. By execution of this Consent Judgment, Settling Defendants do not

1 admit any violations of Proposition 65 or the Unfair Competition Law or any other law and
2 specifically denies that it has committed any such violations and maintains that all Wine
3 Vinegar products it has sold and distributed in California have been and are in compliance
4 with all laws. Nothing in this Consent Judgment shall be construed as an admission by
5 Settling Defendants of any fact, finding, conclusion, issue of law, or violation of law.
6 However, this paragraph shall not diminish or affect the responsibilities and duties of the
7 parties under this Consent Judgment.

8 1.7 For the purposes of this Consent Judgment, the term "Effective Date" shall
9 mean the date upon which this Consent Judgment is approved and entered as a Judgment by
10 the Court.

11 2. CLEAR AND REASONABLE WARNINGS

12 2.1 Warning Standard. Settling Defendants shall not sell or offer for sale in
13 California Wine Vinegars that contain lead at levels that exceed 34 parts per billion ("ppb")
14 unless warnings are given in accordance with one or more of the provisions set forth below.

15 a. Shelf Warning. Settling Defendants may provide warning by placing a
16 notice on the top shelf of any rack of shelves in Settling Defendants' stores where Wine
17 Vinegars are sold. The warning shall state as follows: "CALIFORNIA PROPOSITION 65
18 WARNING: The Red Wine Vinegars and Balsamic Vinegars on these shelves contain lead,
19 a chemical known to the State of California to cause birth defects and other reproductive
20 harm." Each sign shall be no smaller than 5 inches x 7 inches, and the form and type shall
21 be substantially similar to that which is attached hereto as Exhibit A.

22 b. Product Labeling. A warning may be placed on the packing, labeling
23 or directly onto all Red Wine Vinegar products that includes the language as follows:
24 "WARNING: This product contains lead, a chemical known to the State of California to
25 cause birth defects and other reproductive harm." Product label warnings shall be placed
26 with such conspicuousness as compared with other words, statements, designs and/or
27 devices as to render it likely to be read and understood by an ordinary individual under
28 customary conditions of use or purchase.

1 2.4 Testing shall be conducted by a testing laboratory with Environmental
2 Laboratory Certification from the State of California, Department of Health Services,
3 Environmental Laboratory Accreditation Program. Settling Defendant may rely on those test
4 results so long as the facility that performed the tests confirms in writing that it utilized the
5 testing protocol of Professor A. Russell Flegal, attached hereto as Exhibit B. As used in this
6 Consent Judgment "less than 34 ppb" means that 10 samples of each individual product have
7 been tested in accordance with the requirements set forth in this Consent Judgment and that
8 the raw results from the ten (10) samples tested have a lead concentration with an arithmetic
9 mean of less than 34 parts per billion lead and no more than one sample exceeding 50 parts
10 per billion lead , regardless of the source of the lead.
11

12 a. At least 60 days before any proposed discontinuance of any warnings
13 pursuant to this paragraph, Settling Defendant proposing such discontinuance shall provide
14 to ELF the results, the underlying raw data, and a description of the test methodology used.
15 ELF shall keep all such information confidential except as is necessary to contest the
16 exemption from warning of the product. Should ELF dispute for any reason the
17 discontinuance of any warning, the dispute may be submitted by either party to the Court for
18 resolution on motion. Unless and until such motion is resolved favorably to Settling
19 Defendant, the warning in question may not be discontinued. If there is no objection or the
20 objection is resolved favorably to the Settling Defendant, the subject product that tests less
21 than 34 ppb shall not bear a warning label under paragraph 2.1(b) nor placed on shelf
22 referenced by a shelf sign under paragraph 2.1(a).
23

24 b. Nothing in this Consent Judgment shall require any Settling Defendant or
25
26
27
28

1 supplier of Wine Vinegar to conduct any testing of any such vinegar.
2

3 2.5 Provisions of the Warning in paragraphs 2.1 or 2.2 of this Consent Judgment
4 shall fully and completely satisfy Settling Defendants' obligations to provide a warning for
5 all Wine Vinegars with respect to the presence of lead under Proposition 65, the California
6 Business and Professions Code, and all federal, state or local laws, regulations, or
7 ordinances.
8

9 2.6 If ELF settles this, or any lawsuit regarding the same allegations as in the
10 instant Complaint, wherein any retailer is permitted to provide a warning regarding lead in
11 Wine Vinegar that is different in content, method or appearance, each Settling Defendant,
12 shall, at its discretion, have the option to warn in the manner alleged in section 2.1, or in the
13 manner by the subsequent settlement. Settling Defendants shall have the warnings placed no
14 later than sixty (60) days after entry of this Consent Judgment.
15
16

17 3. MONETARY RELIEF

18 3.1 Berberian shall pay to ELF the sum of \$2,000 to be applied toward
19 its costs, attorneys' fees and a *cypres* donation. The distribution of the funds shall be at the
20 sole discretion of ELF. The settlement draft shall be delivered to one of ELF's counsel,
21 Alan M. Caplan, Bushnell, Caplan & Fielding, LLP, 221 Pine Street, Suite 600, San
22 Francisco, California 94104, within five (5) business days after the entry of this Consent
23 Judgment. These Settlement Proceeds shall be delivered to ELF's counsel, and ELF shall
24 have the sole and exclusive responsibility of apportioning and paying to the State of
25 California any portion of the Settlement Proceeds as required by California Health & Safety
26 Code § 25249.12(d), and Berberian shall have no liability if payments to the State of
27
28

1 California are not made by ELF.

2
3 3.2 Bristol Farms shall pay to ELF the sum of \$8,000 to be applied toward
4 its costs, attorneys' fees and a *cypres* donation. The distribution of the funds shall be at the
5 sole discretion of ELF. The settlement draft shall be delivered to one of ELF's counsel,
6 Alan M. Caplan, Bushnell, Caplan & Fielding, LLP, 221 Pine Street, Suite 600, San
7 Francisco, California 94104, within five (5) business days after the entry of this Consent
8 Judgment. These Settlement Proceeds shall be delivered to ELF's counsel, and ELF shall
9 have the sole and exclusive responsibility of apportioning and paying to the State of
10 California any portion of the Settlement Proceeds as required by California Health & Safety
11 Code § 25249.12(d), and Bristol Farms shall have no liability if payments to the State of
12 California are not made by ELF.
13
14

15
16 3.3 These payment shall be the only monetary obligations of the Settling
17 Defendants with respect to this Consent Judgment; each party shall bear its own attorneys'
18 fees and costs.

19
20 3.4 ELF agrees to comply with the reporting requirements referenced in California
21 Health & Safety Code § 25249.7(f). Pursuant to the regulations promulgated under that
22 section, ELF shall present this Consent Judgment to the California Attorney General's
23 Office within two (2) days after receipt of all necessary signatures. ELF also agrees to serve
24 a copy of the Noticed motion to approve and enter the Consent Judgment on the Attorney
25 General's Office at least forty-five (45) days prior to the date set for hearing of the motion in
26 the Superior Court of the City and County of San Francisco.
27
28

3.5 The Settling Parties acknowledge that, pursuant to Health & Safety Code

1 § 25249.7, a noticed motion must be filed to obtain judicial approval of the Consent
2 Judgment. Accordingly, the Settling Parties agree to file a joint motion for approval of the
3 settlement, which shall be prepared by ELF within a reasonable period of time after the date
4 this agreement is signed by all parties.
5

6 4. MODIFICATION OF CONSENT JUDGMENT
7

8 4.1 This Consent Judgment may be modified by written agreement between ELF
9 and the Settling Defendant(s), after noticed motion, and upon entry of a modified Consent
10 Judgment by the Court thereon, or upon motion of ELF or the Settling Defendant as
11 provided by law or upon entry of a modified Consent Judgment by the Court.
12

13 5. APPLICATION OF CONSENT JUDGMENT
14

15 5.1 This Consent Judgment shall apply to and be binding upon ELF and the
16 Settling Defendants, their divisions, subdivisions, parent entities or subsidiaries, and
17 successors or assigns of either of them. officers, directors, and shareholders.
18

19 5.2 Each signatory to this Consent Judgment certifies that he or she is fully
20 authorized by the party that he or she represents to enter into and execute the Consent
21 Judgment on behalf of the party represented and legally bind that party.
22

23 6. CLAIMS COVERED
24

25 6.1 This Consent Judgment is a final and binding resolution between ELF and the
26 Settling Defendants, of any violation of Proposition 65 and Business and Professions Code
27 section 17200, *et seq.*, or any other statutory or common law claim that could have been
28 asserted against the Settling Defendants for failure to provide clear, reasonable and lawful
warnings of exposures to lead that result from the ingestion of Wine Vinegar.

1 6.2 For purposes of this Consent Judgment, the term "Wine Vinegar" shall mean
2 any red vinegar, including but not limited to balsamic vinegar, that contains wine as a
3 constituent. Nothing in this section shall be construed to affect the liability of any defendant
4 in this Action other than the Settling Defendant.
5

6 6.3. Release of Settling Defendant. In further consideration of the promises and
7 agreements herein contained, and for the payments to be made pursuant to Paragraph 3.1,
8 ELF, on behalf of itself, its past and current agents, representatives, attorneys, successors
9 and/or assignees, and in the interest of the general public, hereby waives all rights to
10 institute or participate in, directly or indirectly, any form of legal action and releases all
11 claims, including, without limitation, all actions, causes of action, in law or in equity, suits,
12 liabilities, demands, obligations, damages, costs, fines penalties, losses or expenses,
13 including, but not limited to, investigation fees, expert fees and attorneys' fees of any nature
14 whatsoever, whether known or unknown, fixed or contingent against the Settling Defendants
15 and each of their customers, owners, parent companies, corporate affiliates, subsidiaries and
16 its respective officers, directors, attorneys, representatives, shareholders, agents, and
17 employees arising under Proposition 65, Business and & Professions Code § 17200, *et seq*
18 and Business & Professions Code § 17500, *et seq.*, related to the Settling Defendants'
19 alleged failure to warn about exposures to or identification of lead contained in Wine
20 Vinegars.
21

22 ELF and the Settling Defendants further agrees and acknowledges that this Consent
23 Judgment is a full, final, and binding, resolution of any violations of Proposition 65,
24 Business & Professions Code § 17200, *et seq.* and Business & Professions Code § 17500, *et*
25

1 *seq.*, that have been or could have been asserted in the Complaint against the Settling
2 Defendants for their alleged failure to provide clear and reasonable warnings of exposure to
3 or identification of lead contained in Wine Vinegars.
4

5 In addition, ELF, on behalf of its, itself, attorneys and its agents, waives all rights to
6 institute or participate in, directly or indirectly, any form of legal action and releases all
7 claims against the Settling Defendants arising under Proposition 65, Business & Professions
8 Code § 17200, *et seq* and Business & Professions Code § 17500, *et seq.*, related to the
9 Settling Defendants' alleged failures to warn about exposures to or identification of lead
10 contained in the Wine Vinegars and for all actions or statements regarding the alleged
11 failures to warn about exposures to or identification of lead contained in the Wine Vinegars
12 made by Settling Defendants or their attorneys or representatives, in the course of
13 responding to those alleged violations of Proposition 65, Business & Professions Code §
14 17200, or Business & Professions Code § 17500, as alleged in the Complaint.
15
16
17

18 It is specifically understood and agreed that ELF and the Settling Defendants intend
19 that Settling Defendants' compliance with the terms of this Consent Judgment will resolve
20 all issues and liability, now and in the future, concerning the Settling Defendants' alleged
21 violation of the requirements of Proposition 65, Business & Professions Code § 17200, *et*
22 *seq.* and Business & Professions Code § 17500, *et seq.*, as to lead in Wine Vinegars.
23
24

25 6.4 Release of ELF. Settling Defendants waive all rights to institute any form of
26 legal action against ELF or its attorneys or representatives, for all actions taken or statements
27 made by ELF and its attorneys or representatives, in the course of seeking enforcement of
28 Proposition 65, Business & Professions Code § 17200, *et seq.* or Business & Professions

1 Code § 17500, *et seq.*, in these Actions.

2
3 7. RETENTION OF JURISDICTION

4 7.1 This Court shall retain jurisdiction of this matter to implement this Consent
5 Judgment.

6
7 8. COURT APPROVAL

8 8.1 If this Consent Judgment is not approved by the Court, it shall be of no force
9 or effect and cannot be used in any proceeding for any purpose.

10 9. ENFORCEMENT OF CONSENT JUDGMENT WITH REGARD
11 TO RETAIL STORES IN CALIFORNIA

12 9.1 Before moving to enforce the terms and conditions of this Consent Judgment
13 against a Settling Defendant with respect to an alleged violation occurring at a retail store
14 located in California, ELF must follow the procedures set forth in subsections 9.2 through
15 9.4.

16
17 9.2 In the event that ELF and/or its attorneys, agents or assigns, identify one or
18 more retail stores in California owned and operated by a Settling Defendant at which Wine
19 Vinegars are sold (hereinafter "retail outlet") for which the warnings required under
20 paragraph 2 of this Consent Judgment are not being given, ELF shall notify, in writing, that
21 Settling Defendant of such alleged failure to warn (the "Notice of Breach"). The Notice of
22 Breach shall be sent by first-class mail, with proof of service within sixty (60) days of the
23 date the alleged violation was observed. The Notice of Breach shall identify the date the
24 alleged violation was observed and the retail outlet in question, and reasonably describe the
25 nature of the alleged violation with sufficient detail to allow the Settling Defendant to
26
27
28

1 determine the basis of the claim being asserted and the identities of the Wine Vinegars to
2 which those assertions apply.
3

4 9.3 In the event that ELF identifies a specific retail outlet, other than the specific
5 one identified in subsection 9.2 of this Consent Judgment, not giving warnings for Wine
6 Vinegars as required under paragraph 2, ELF shall serve that Settling Defendant with
7 another Notice of Breach in the manner described in subsection 9.2 and provide the same
8 information as required in subsection 9.2.
9

10 9.4 ELF shall take no further action against that Settling Defendant unless ELF
11 discovers, at least thirty (30) days after service of the Notices of Breach served pursuant to
12 subsections 9.2 and 9.3, another failure to warn for any Wine Vinegars at the same retail
13 outlet(s) identified in the Notices of Breach served pursuant to subsections 9.2 and 9.3.
14

15 10. GOVERNING LAW
16

17 10.1 The terms of this Consent Judgment shall be governed by the laws of the State
18 of California. In the event that Proposition 65 is repealed or is otherwise rendered
19 inapplicable by reason of law generally, or as to Wine Vinegars specifically, then the
20 Settling Defendants shall have no further obligations pursuant to this Consent Judgment
21 with respect to, and to the extent those Wine Vinegars are so affected.
22

23 11. EXCHANGE IN COUNTERPARTS
24

25 11.1 Stipulations to this Consent Judgment may be executed in counterparts by
26 and/or facsimile which taken together shall be deemed to constitute one document.
27

28 ///

///

1 12. NOTICES

2
3 12.1 All correspondence and notices required to be provided pursuant to this
4 Consent Judgment shall be in writing and personally delivered or sent by: (1) first-class,
5 registered, certified mail, return receipt requested, or (2) overnight courier on ELF or that
6 Settling Defendant by the others at the addresses listed in Exhibit C. Either ELF or Settling
7 Defendants may specify a change of address to which all notices and other communication
8 shall be sent.
9

10 IT SO STIPULATED:

11
12
13 DATED: 7/25/06

ENVIRONMENTAL LAW FOUNDATION

14
15
16 By: 
17 JAMES WHEATON

18 DATED: 7/24/06

BERBERIAN ENTERPRISES, INC.

19
20
21 By: 
22 FR. V. P.

23
24
25 DATED: _____

BRISTOL FARMS

26
27
28 By: _____

1 12. NOTICES

2
3 12.1 All correspondence and notices required to be provided pursuant to this
4 Consent Judgment shall be in writing and personally delivered or sent by: (1) first-class,
5 registered, certified mail, return receipt requested, or (2) overnight courier on ELF or that
6 Settling Defendant by the others at the addresses listed in Exhibit C. Either ELF or Settling
7 Defendants may specify a change of address to which all notices and other communications
8 shall be sent.
9

10 IT SO STIPULATED:

11
12
13 DATED: _____

ENVIRONMENTAL LAW FOUNDATION

14
15
16 By: _____
17 JAMES WHEATON

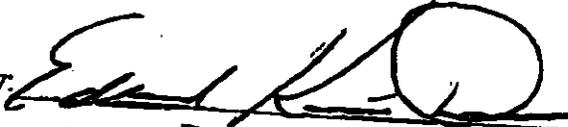
18 DATED: _____

19 BERBERIAN ENTERPRISES, INC.

20
21 By: _____
22 _____

23
24
25 DATED: _____

26 BRISTOL FARMS

27
28 By: 
Edmund R. [unclear], Chairman & CEO

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28

IT IS SO ORDERED, ADJUDGED AND DECREED:

DATED: _____

JUDGE OF THE SUPERIOR COURT

EXHIBIT A

**CALIFORNIA
PROPOSITION 65
WARNING:**

**! The Red Wine Vinegars and
Balsamic Vinegars on these
shelves contain lead, a chemical
known to the State of California
to cause birth defects and other
reproductive harm.**

EXHIBIT B

Determination of lead in vinegar by ICP-MS and GFAAS: evaluation of different sample preparation procedures

Kuria Ndung'u^{a,b,*}, Sharon Hibdon^a, A. Russell Flegal^a

^a Environmental Toxicology, UCSC University of California, Santa Cruz, CA 95064, USA

^b Institute of Applied Environmental Research (ITM), Stockholm University, Frescatiweg 54, S-106 91 Stockholm, Sweden

Received 4 November 2003; received in revised form 12 February 2004; accepted 12 February 2004

Available online 17 April 2004

Abstract

Lead concentrations of 39 different types of vinegars ($15\text{--}307\ \mu\text{g l}^{-1}$ in balsamic vinegars and $36\text{--}50\ \mu\text{g l}^{-1}$ in wine vinegars) were determined using both inductively coupled plasma mass spectrometry (ICP-MS) and graphite furnace atomic absorption spectrometry (GFAAS). Although the precision of direct analyses, following simple aqueous dilutions, with either instrumental method was poor, that precision, following nitric acid and/or hydrogen peroxide digestions, markedly improved with either instrument and the values obtained with the two instruments were in good agreement. The efficacy of different digestions, including (1) nitric acid using a heating block, with or without addition of hydrogen peroxide and (2) mixtures of nitric acid and hydrogen peroxide using ultraviolet (UV) photolysis, were then assessed. The latter procedure was found to be much faster and more efficient, but it was limited by the relatively high levels of contaminant lead in hydrogen peroxide. Consequently, it is recommended that lead concentrations in vinegar be measured following a nitric acid digestion and UV photolysis to oxidize all organic matter before ICP-MS or GFAAS analysis; and it is further recommended that the thermal settings for the latter analyses be adjusted to account for the apparent presence of relatively volatile organolead compounds in vinegar digesta.

© 2004 Elsevier B.V. All rights reserved.

Keywords: Organolead compounds; Vinegar; ICP-MS

1. Introduction

Exposure to contaminant lead remains a public concern because of its pervasiveness in the environment and increasing evidence of lead's sub-lethal toxicities at exposure levels lower than previously thought harmful [1]. In response to those concerns, there have been orders of magnitude reductions in atmospheric emissions of industrial lead, which have resulted in a pronounced decrease in blood lead levels in the US and elsewhere [2]. Now, the most common route of exposure to the general population, in countries where leaded gasoline has been banned, is through the ingestion of food and water contaminated with lead [3].

Among those foods is vinegar, which can contain relatively high levels of lead [4,5]. It may, like wine, come from the grapes vinegar is made from and it might be of

endogenous or anthropogenic origin [6,7]. Conversely, the lead may come from contamination during the vinegar production process [8].

Although there are numerous published studies on the concentration of lead in wine, only a handful of studies have looked at the concentration of lead in vinegar [4,5,9,10]. While some of those studies measured the lead in vinegar or wine directly after simple dilution [10–12], quite often a sample clean-up step was employed prior to the instrumental analysis. This pretreatment is often needed because, in addition to acetic acid and alcohol, both vinegar and wine contain suspended particles and polymeric organic compounds, particularly sugars, which interfere with GFAAS and ICP-MS measurements. The polymeric organic matter might cause blockage of the injector tube and cones of the ICP, due to incomplete pyrolysis of the sugars in the plasma and formation of residual carbon deposits [13]. During the GFAAS analysis, incomplete pyrolysis of the organic matter produces fumes and accumulation of carbonaceous residue

* Corresponding author. Tel: +46-86747236; fax: +46-86747636.
E-mail address: kuria.ndungu@im.uu.se (K. Ndung'u).

after several graphite tube firings which adversely affect the analysis [11].

Two types of oxidation are most common: acidification and irradiation. Wet digestion using nitric acid is usually employed to oxidize the organic matter, and those oxidative digestions are often accelerated by heating the samples in Teflon or other inert and trace metal clean containers on a heating block or heating plate. The addition of hydrogen peroxide also speeds up the oxidation process, but most peroxides contain relatively high amounts of lead. Alternatively, ultraviolet (UV) and/or microwave energy have also been used to oxidize the organic matter in wine [13,14] which is a precursor of many vinegars. Since UV photolysis has not previously been applied to vinegar digestions, and the relative accuracy and efficacy of the different analytical methods for measuring lead in vinegar have not been previously determined.

2. Background

Vinegar is produced by a two-stage fermentation process of suitable sugar or starch containing agricultural material such as grapes, apples, rice, garlic or even onions [15]. Besides vinegar from red and white wine, there are special products such as vinegar from Jerez (Sherry vinegar) in Spain or balsamic vinegar elaborated from a specific region of Italy, Modena [15] Aceto Balsamico di Modena, a typical Italian product is produced from fresh grape must, which is concentrated up to a third of its original volume by a slow heating process. The traditional method of production requires storage in different wood barrels up to 25 years. Another balsamic vinegar is produced by blending the concentrated must with acetic acid, and the mixture is allowed to mature in wooden barrels to develop the typical organoleptic properties [15].

Consequently, there may be pronounced differences in the organic composition of different types of vinegars, including different balsamic vinegars. There may also be large differences in the lead concentrations of different vinegars, based on the origins of the ingredients and the production process. Both of these variables complicate accurate and precise measurements of lead in vinegar.

3. Experimental

3.1. Reagents

All solutions were prepared with de-ionized water ($18 \text{ M}\Omega \text{ cm}^{-1}$) from a Milli-Q[®] analytical reagent-grade water purification system (Millipore, Bedford, MA). Calibration standard solutions and internal standards were prepared from commercial lead standard solution (Spex Plasma, Edison, NJ). Trace metal grade (TMG) nitric acid and hydrochloric acid (Fisher Scientific, Pittsburgh, PA) were used for cleaning laboratory ware. Optima grade nitric acid (Fisher) was used for the preparation of calibration standard solutions and analytical solutions. High purity hydrogen peroxide 30% (Ultrapur, Bayer, Pittsburg, NJ), together with nitric acid was used for both heat and UV digestions. The matrix modifier used for GFAAS analysis contained 0.05 mg of $\text{NH}_4\text{H}_2\text{PO}_4$ and 0.003 mg of $\text{Mg}(\text{NO}_3)_2$ per 5 μl of solution (Environmental Express, Mt. Pleasant, SC).

3.2. Instrumentation

3.2.1. ICP-MS

All ICP-MS measurements were made with a Thermo-Finnigan Element magnetic sector high resolution ICP-MS using a Glass Expansion Conical nebulizer, a Scott-type double pass spray chamber (cooled to 10°C) and standard nickel cones. Since there were small or no polyatomic interferences for lead, it was analyzed at low resolution ($r = 300$) using ^{209}Bi as an internal standard. The instrument operating parameters and data acquisition details are listed in Table 1.

3.2.2. GFAAS

Graphite furnace atomic absorption spectroscopy (GFAAS) analyses were made on a Perkin-Elmer SIMAA 6000 instrument, fitted with a Zeeman background corrector and AS72 auto sampler. End capped, transversely heated pyrocoated graphite tubes with an integrated L'vov platform (Perkin-Elmer) were used. A lead electrodeless discharge lamp (Perkin-Elmer) was used at the recommended line of 283.3 nm and a lamp current of 450 mA. Magnesium nitrate ($\text{Mg}(\text{NO}_3)_2$)/ammonium phosphate ($\text{NH}_4\text{H}_2\text{PO}_4$) was

Table 1
ICP-MS operating and acquisition parameters

RF power (W)	1250
Plasma gas flow (l min^{-1})	13
Auxiliary gas flow (l min^{-1})	0.75
Nebulizer gas flow (l min^{-1})	0.85–0.95 (optimized daily)
Sample flow rate ($\mu\text{l min}^{-1}$)	60

Data acquisition (low resolution, 200 scans)

Isotope	% mass window	Sample time (s)	Samples/peak	Segment duration (s)	Detection mode
^{208}Pb	5	0.001	100	0.050	Count
^{209}Bi	5	0.001	100	0.050	Count

Table 2
Optimized GFAAS program for measuring lead concentrations ($\mu\text{g l}^{-1}$) in vinegar, following acid, UV and/or microwave digestion

Temperature (°C)	Ramp time (s)	Hold time (s)	Gas flow (ml min^{-1})	Read
130	5	30	250	No
130	15	30	250	No
700	15	30	250	No
1400	0	3	0	Yes
2450	1	3	250	No

used as a chemical modifier. The optimized, based on tests conducted for this report (see following section on GFAAS Program Optimization) GFAAS program used is shown in Table 2.

3.3. Samples

Vinegar samples were purchased from retail stores in California. Fifty-two different types of balsamic vinegar, four wine vinegars, one apple cider vinegar, one rice vinegar and one garlic vinegar were analyzed. The vinegars were mostly in glass bottles, but some were in plastic or ceramic bottles.

3.4. Contamination control

The exteriors of the bottles were rinsed with deionized water before opening in a HEPA filtered (Class 100) trace metal clean laboratory. Aliquots were placed in Teflon digestion vessels that were cleaned with Micro-90 liquid laboratory grade detergent (Cole-Parmer, Vernon Hills, IL) and deionized water when first used or after an incomplete digestion. Subsequently, digestion vessels were re-cleaned by soaking them overnight in 8 M TMG hydrochloric acid followed by at least 3 h in hot TMG nitric acid. The vessels were then rinsed with reagent water and dried under class 100 HEPA-filtered laminar flow air. All other plastic ware (polyethylene or Teflon) used for storing analytical solutions were cleaned using the same procedure, dried, capped, and stored under class 100 HEPA-filtered laminar flow air or double bagged in trace metal clean, self-locking (Zip loc®) plastic bags. The GFAAS was in a HEPA-filtered air room and directly beneath a HEPA-filtered (Class 100) laminar flow canopy within a plastic enclosure.

3.5. Vinegar digestions

3.5.1. Heating block digestions

Analytical portions were weighed (0.5–1.0 g) into Teflon digestion vessels, and 10 ml of TMG nitric acid was added. Vessels were covered loosely with acid cleaned Teflon lids and placed in the heating block (CPI International, Santa Rosa, CA). They were initially digested at 50 °C for 2–3 h to avoid sputtering then the temperature was increased to 90 °C, and then digested to dryness. After cooling, the digests were dissolved in 1 M TMG nitric acid, producing a clear to light

yellow analytical solutions. These were then analyzed for their lead concentration by GFAAS or ICP-MS.

3.5.2. UV digestions

The UV digestion unit consisted of a medium pressure mercury vapor discharge tube (1200 W; Hamova, Union, NJ) positioned on the ceiling of a purpose-built aluminum housing, (36 cm × 29 cm × 23 cm; UVO-cleaner model 342, Jelight Inc., Laguna Hills, CA), which was cooled by a fan. A digital photometer (model JL1400A, Jelight Inc., Irvine, CA) was used to monitor the power of the UV radiation during the oxidation ($x = 9.2 \pm 0.4 \text{ mW cm}^{-2}$ during the continuous operation of the Hg lamp).

The digestions were carried out by placing 16 custom-made PTFE 15 ml digestion cups fitted with quartz glass caps in the UV digestion unit.

Vinegar samples (0.5 g) were weighed in tared Teflon vials. These and 1 ml of TMG nitric acid and 0.5 ml of 30% hydrogen peroxide were added prior to the UV treatment.

3.6. Quality control

Sample batches consisted of 24 analytical portions including several duplicate samples. Spikes of lead were added (90–150 $\mu\text{g l}^{-1}$) prior to digestion to several vinegar analytical portions representative of the variety of products. Standard solutions were analyzed after every 10 analytical solutions to ensure instrument performance. Each analytical batch contained at least three method blanks, three spiked analytical samples, and three reference materials. Because there is no commercially available certified reference material for lead in vinegar (or wine), we used the National Institute of Standards and Technology (NIST) 1640 Standard Reference Material (SRM) for trace metals in natural waters (NIST, Gaithersburg, MD) with a lead concentration (where X is the mean \pm S.D.) of $27.89 \pm 0.14 \mu\text{g l}^{-1}$ to monitor the extraction efficiency of the digestion process.

4. Results and discussion

4.1. Nitric acid digestion

As previously noted, only a small number of studies have been published on the determination of lead in vinegar [4,5,9,10]. Most of them have employed a sample pretreatment to destroy the organic matter, which might interfere with GFAAS or ICP-MS analyses. In contrast, a few studies have reported direct analysis of lead in wine by GFAAS [16] or ICP-MS [11,17] after a simple aqueous dilution.

However, our attempts to analyze vinegar with or without dilution by either GFAAS or ICP-MS resulted in erroneously high lead concentration values (compared to nitric acid digested vinegar) and relatively poor precision. This analytical variability is illustrated in Table 3, which is a summary of the lead determination in four different types of balsamic

Table 3.
Comparison of lead concentrations in four different balsamic vinegars analyzed by GFAAS and ICP-MS with and without nitric acid digestion

Vinegar	Lead concentration ^a ($\mu\text{g l}^{-1}$)			
	Simple digestion		Digested with nitric acid	
	GFAAS	ICP-MS	GFAAS	ICP-MS
Balsamic-1	595 (18)	447 (7)	319 (9)	306 (6)
Balsamic-2	653 (14)	286 (9)	198 (7)	174 (2)
Balsamic-3	277 (34)	68 (14)	61 (7)	68 (5)
Balsamic-4	348 (4)	188 (17)	99 (9)	95 (4)

^a Mean and relative standard deviation (values in parenthesis) of at least six determinations.

vinegars (six replicate digestions or analyses). Because of their complex organic content, those vinegars proved to be the most difficult to analyze by either GFAAS or ICP-MS and with and without a prior nitric acid digestion.

Specifically, measurements with both types of instruments yielded significantly ($P \leq 0.05$, paired t test) higher lead concentrations in balsamic vinegars after simple aqueous dilutions compared to measurements after acid digestions. The disparity was greater in direct analyses of undigested diluted vinegars by GFAAS. In addition to vinegar matrix interferences, we noticed irreproducible sample deposition on the graphite tube due to adhesion of vinegar solutions to the Teflon GFAAS deposition tubing. Moreover, this problem persisted after filtering and diluting the vinegars.

The agreement and precision of the analyses between the two instruments was greatly improved ($R = 0.997$, $n = 0.94$, simple linear regression) after nitric acid digestions. These improvements are attributed to the oxidation of organic matter. That destruction eliminates interferences resulting from nonspecific absorption and scattering of light due to concomitant species in the vinegar solutions.

4.2. UV and heat digestion

Nitric acid, and to a lesser extent hydrogen peroxide, are widely used for wet digestions of organic and inorganic matter prior to instrumental analyses of metals. The oxidative digestions are accelerated by heating the samples in Teflon or other inert, trace metal clean containers on a heating block or heating plate. Those thermal energy sources are now often being replaced by microwave and UV radiation in sample preparations where acid digestion is necessary, because they may be faster and may be done within a closed system [13,18].

Comparing the two methods, nitric acid digestions with UV radiation were faster than those with heating blocks and the digestions were more complete. While the addition of hydrogen peroxide further enhanced the degradation of organic material in the vinegars, the amount of contaminant lead in TMG hydrogen peroxide we used was relatively high ($\sim 15 \mu\text{g l}^{-1}$) and comparable to the lead concentration in some of the vinegars. Thus, cleaner hydrogen peroxide is

necessary for digestion of vinegars with lead concentration in the low to sub $\mu\text{g l}^{-1}$ level.

4.3. GFAAS analysis

Although the instrument manufacturer (Perkin-Elmer) recommended a maximum ashing and atomization temperatures of 400 and 1400 °C, respectively, in the furnace program for lead determination, the use of chemical modifiers allows much higher ashing and atomization temperatures. Preschi et al. [11] used an ashing temperature of 1000 °C and an atomization temperature of 1800 °C to determine lead in diluted wine samples and nitric acid wine digests using a phosphate/magnesium matrix modifier. Buldini et al. [19] also used a phosphate/magnesium modifier and were able to determine lead in nitric acid wine digests using ashing and atomization temperatures of 900 and 1800 °C, respectively.

In the absence of a vinegar or similar matrix SRM with certified lead concentration, we initially started the optimization of the furnace program using digested vinegar spikes and NIST SRM 1640 (natural water) that had undergone a similar nitric acid digestion process as the vinegars. We used the manufacturer's recommended ashing and atomization temperatures with a $\text{Mg}(\text{NO}_3)_2/\text{NH}_4\text{H}_2\text{PO}_4$ chemical modifier. We found ashing and atomization temperatures of 800 and 1400 °C, respectively, to be optimum for analysis of digested natural water SRM and quantitative recovery. However, the same furnace program produced low lead recoveries of spiked digested vinegar samples.

An investigation of the GFAAS measurements of vinegar digests with similar lead concentrations as the SRM showed a sharp drop in absorbance between 700 and 800 °C of the digested vinegar samples, but not for the SRM. This disparity is shown in Fig. 1. It contains plots of the variation of absorbance during ashing and atomization temperatures steps of the two types of samples.

Curvatures in both plots indicate the digestion of the vinegar samples produced a relatively labile lead compound(s). Their volatilization between 700 and 800 °C markedly altered the measurements of lead concentrations of the vinegar, which was not replicated in the measurements of lead in the SRM. This thermal variability underscores the importance of close investigation of the furnace program optimization for different sample types and matrices.

4.4. Quality control

Process blanks (reagent water) were also analyzed together with the samples. The mean blank lead concentration was $0.03 \mu\text{g l}^{-1}$ ($n = 4$) with a standard deviation of 0.04, giving a detection limit of $0.12 \mu\text{g l}^{-1}$ analyzed by GFAAS after nitric acid and heat digestion. The spike recovery ($x \pm \text{S.D.}$) of six different vinegars was $96 \pm 5\%$, while the mean recovery of NIST 1640 SRM digests was $97.4 \pm 1.3\%$. The relative standard deviation for duplicate analysis was $< 8\%$.

to process the vinegar and increase the oxidation of its organic constituents. Although the digestion times may be further enhanced with the addition of hydrogen peroxide, the amount of contaminant lead in TMG hydrogen peroxide is too high for measurements of lead concentrations in vinegars with concentrations $<50 \mu\text{g l}^{-1}$. Therefore, we recommend nitric acid digestion of vinegars before ICP-MS or GFAAS determination, and that the latter measurements use ashing and atomization temperatures of 600 and 1300 °C, respectively, rather than the manufacturer's recommended settings because of the apparent volatilization of relatively labile forms of lead in vinegars above those temperatures.

Acknowledgements

We are grateful to Ann Gonzalez for help with vinegar digestion and Rob Finkle for help with the ICP-MS analysis. This research was funded by Environmental Law Foundation.

References

- [1] K.D. Rasmussen, A. Shun, Z.H. Lee, I. Gordner, *J. Occup. Environ. Med.* 45 (2003) 546.
- [2] CDC, Second National Report of Human Exposure to Environmental Chemicals (2002), CDC, Center for Disease Control and Prevention, 2003, <http://www.cdc.gov/exposurereport/metals/>, accessed on 6 October 2003.
- [3] CDC, Toxicological Profile Information: Toxic Profile for Lead, CDC, Center for Disease Control and Prevention, Agency for Toxic Substances and Disease Registry (ATSDR), 1999, <http://www.atsdr.cdc.gov/toxprofiles/>, accessed on 6 October 2003.
- [4] F. Carradini, L. Marcheselli, A. Marchetti, C. Fusi, C. Blamirelli, *J. Assoc. Int.* 77 (1994) 714.
- [5] A. Acosta, C. Diaz, A. Heruleson, D. Gonzalez, *Environ. Contam. Tox.* 51 (1993) 832.
- [6] V.R. Angelova, A.R. Ivanov, D.M. Bralov, *J. Sci. Food Agric.* 79 (1999) 713.
- [7] V. Oroszova, A. Kozmar, A. Kufic, V. Velkovic, *J. Trace Microprobe Tech.* 21 (2003) 171.
- [8] M.L. Guzman, C. Harco-Paglal, A.M. Cansino, A.M. Troncoso, A.G. Gonzalez, *Talanta* 45 (1997) 379.
- [9] A. Del Signore, B. Campia, F. Di Giacomo, *J. Assoc. Int.* 81 (1996) 1087.
- [10] Z.J. Sutarovic, N.J. Marjanovic, N.M. Dostanic, *Nahrung* 41 (1997) 111.
- [11] G.P.G. Fraschi, C.R. Dakshika, M. de Menna, J.A. Nobrega, J.A.G. Niza, *Spectrochim. Acta B* 56 (2001) 1987.
- [12] A.M.T. Gonzalez, M.G. Chona, *Nahrung* 32 (1988) 743.
- [13] C.M.R. Almeida, M. Vasconcelos, *J. Anal. Atom. Spectrom.* 14 (1999) 1815.
- [14] C.R. Quedt, S.M. Nolas, L. Van Navel, I. Papadakis, P.D.P. Taylor, *J. Anal. Atom. Spectrom.* 16 (2001) 1091.
- [15] W. Tuckaya, M.L. Morales, M.C. Garcia-Panilla, A.M. Troncoso, *Trends Food Sci. Technol.* 13 (2002) 12.
- [16] Z.Y. Zuo, M. Zhang, Z.A. Sun, D.R. Wang, *Spectrosc. Spectr. Anal.* 22 (2002) 839.
- [17] C.M.R. Almeida, M. Vasconcelos, M. Barbato, B. Medina, *Anal. Bioanal. Chem.* 374 (2002) 314.
- [18] Q.H. Ho, F. Liang, H.Q. Zhang, L.W. Zhao, Y.K. Huan, D.Q. Song, *Trace Trends Anal. Chem.* 18 (1999) 479.
- [19] P.L. Beldick, S. Cavall, J.L. Sharma, *J. Agric. Food Chem.* 47 (1999) 1993.
- [20] C.M.R. Almeida, M. Vasconcelos, *J. Agric. Food Chem.* 51 (2003) 3012.
- [21] J. Kristel, M. Weber, M. Siskovic, *Anal. Bioanal. Chem.* 373 (2002) 208.

EXHIBIT C

1 James R. Wheaton, Esq.
ENVIRONMENTAL LAW FOUNDATION
2 1736 Franklin Street, Ninth Floor
Oakland, CA 94612
3 Tel: (510) 208-4555
Fax: (510) 208-4562
4

5 COUNSEL FOR DEFENDANTS BERBERIAN ENTERPRISES, INC.
AND BRISTOL FARMS

6 Michael J. Nangano, Esq.
STREETER & NANGANO
7 445 South Figueroa Street, 27th Floor
Los Angeles, CA 90071
8 Tel: (213) 612-7716
Fax: (213) 612-7717
9

10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28